# Phenylethanoid Glycosides from Digitalis purpurea and Penstemon linarioides with PKC $\alpha$-Inhibitory Activity 

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In a continuation of our search for potential tumor inhibitors from plants, it was found that the $\mathrm{CH}_{2^{-}}$ $\mathrm{Cl}_{2}-\mathrm{MeOH}$ (1:1) extracts from Digitalis purpurea and Penstemon linarioides both showed PKC $\alpha$-inhibitory bioactivity. Bioassay-directed fractionation of the extract from D. purpurea yielded the new, weakly active phenylethanoid glycoside 2-(3-hydroxy-4-methoxy-phenyl)-ethyl-O-( $\alpha-$-L-rhamnosyl)-(1 $\rightarrow 3$ )-O-( $\alpha$-L-rham-nosyl)-(1 $\rightarrow 6$ )-4-O-E-feruloyl- $\beta$-D-glucopyranoside (1) together with the four known compounds cal ceol ariosideA (2), cal ceol arioside B (3), forsythiaside (4), and plantainosideD (5). The extract from P. Iinarioides yielded the three known glycosides leucosceptoside A (6), acteoside (7), and poliumoside (8), together with the iridoid plantarenaloside (9). All of the isolated compounds, except compound 9, showed inhibitory activity against $\mathrm{PKC} \alpha$ with $\mathrm{IC}_{50}$ values (in $\mu \mathrm{M}$ ) of 125 (1), 0.6 (2), 4.6 (3), 1.9 (4), 14.8 (5), 19.0 (6), 9.3 (7), and 24.4 (8).

As described in the previous paper in this series ${ }^{1}$ we have added a screen for inhibitors of protein kinase C (PKC) to our yeast assays for DNA-damaging agents, ${ }^{2}$ in as much as PKC has emerged as an attractive target for anticancer treatment. ${ }^{3}$ A search of plant extracts for inhibitors of PKC indicated that the detanninated $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (1:1) extracts of both Digitalis purpurea L. (Scrophulariaceae) and Penstemon linarioides Gray (Scrophulariaceae) showed PKC $\alpha$-inhibitory bioactivity, and therefore bioassay-directed fractionation was undertaken on both extracts.
D. purpurea is a well-known herb with a long history of medicinal use. It is the source of the important cardiac glycosides digitoxin, gitoxin, and gitaloxin and has been used medicinally for at least 200 years. ${ }^{4}$ It also contains flavonoid glycosides and an anthraquinone. ${ }^{5}$ Recently, certain phenylethanoids such as desrhamnosyl acteoside, forsythiaside, purpureaside A, and purpureaside B have been isolated from D. purpurea, and it has been reported that acteoside (verbascoside) shows PKC $\alpha$-inhibitory activity. ${ }^{6}$
P. Iinarioides has been reported to contain more than 50 iridoid glycosides, ${ }^{7}$ and various phenylethanoids such as martynoside, ${ }^{8}$ orobanoside, ${ }^{9}$ stansioside, ${ }^{9}$ and acteoside ${ }^{10}$ have been isolated from other plants of this genus. As noted above, acteoside has shown PKC $\alpha$-inhibitory activity and also selectively inhibited aldose reductase and formation of the 5-lipoxygenase product 15-hydroxy-5,8,11,13eicosastetraenoic acid (15-HETE) and $\mathrm{LTB}_{4}$ in human peripheral polymorphonuclear leukocytes, as well as showing antibacterial and cytotoxic activities. ${ }^{11}$

The extracts of both plants were subjected to partition between various organic sol vents and aqueous MeOH , and the bioactivity of each extract against PKC $\alpha$ was concentrated in the $\mathrm{n}-\mathrm{BuOH}$ fraction. For D. purpurea 2.52 g of crude extract gave 1.14 g of active $\mathrm{n}-\mathrm{BuOH}$-soluble material with an $\mathrm{IC}_{50}$ of $11.7 \mu \mathrm{~g} / \mathrm{mL}$ against $\mathrm{PKC} \alpha$, while for P . Iinarioides 2.27 g of crude extract yielded 1.27 g of $\mathrm{n}-\mathrm{BuOH}-$

[^0]soluble material with an $\mathrm{IC}_{50}$ of $17.1 \mu \mathrm{~g} / \mathrm{mL}$ against PKC $\alpha$. Column chromatography of the $\mathrm{n}-\mathrm{BuOH}$ fraction from D . purpurea on Si gel with the solvent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (8:2:0.1) gave seven fractions. The bioactive fraction 6 gave compound $1(4.8 \mathrm{mg})$ and plantainoside (5) ( 8.0 mg ) by preparative TLC on RP-18 plates with the sol vent MeOH $\mathrm{H}_{2} \mathrm{O}(55: 45)$ and repeated column chromatography on Si gel with the sol vent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (8:2:0.1). Using the same conditions, cal ceol ariside A ( $2,16.8 \mathrm{mg}$ ), cal ceolariside $B(\mathbf{3}, 8.8 \mathrm{mg})$, and forsythiaside ( $\mathbf{4}, 22.3 \mathrm{mg}$ ) were isolated from the bioactive fractions 2,3 , and 4 , respectively.
In similar fashion, the $n-B u O H$ fraction from $P$. Iinarioides was subjected to Si gel column chromatography with the sol vents $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (6:1 and 4:1). Fraction 4 yiel ded leucosceptoside A ( $6,120 \mathrm{mg}, 5.2 \%$ ) by further Si gel column chromatography with the solvent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-$ MeOH (6:1). Fraction 6 gave acteoside (7, $145 \mathrm{mg}, 6.4 \%$ ) on purification by polyamide column chromatography, and fraction 8 gave poliumoside ( $8,62.7 \mathrm{mg}, 2.8 \%$ ) and plantarenal oside ( $9,36.8 \mathrm{mg}, 1.6 \%$ ) on Si gel column chromatography with the solvent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (8:2:0.1).

Compound $\mathbf{1}$ had the composition $\mathrm{C}_{35} \mathrm{H}_{50} \mathrm{O}_{19}$ as determined by HRFABMS. Its ${ }^{1} \mathrm{H}$ NMR spectrum, with signals at $\delta 3.88(3 \mathrm{H}, \mathrm{s}), 7.20(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.9 \mathrm{~Hz}), 6.81(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=$ $8.2 \mathrm{~Hz}), 7.08(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=1.9,8.2 \mathrm{~Hz}), 6.38(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=$ 15.9 Hz ) and $7.66(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=15.9 \mathrm{~Hz})$, and its UV spectrum, with $\lambda_{\max }$ at 289 and 330 nm , suggested the presence of a feruloyl moiety, while ${ }^{1} \mathrm{H}$ NMR signals at $\delta$ $6.73(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=2.1 \mathrm{~Hz}), 6.83(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.1 \mathrm{~Hz}), 6.69$ ( 1 H , dd, J = 2.1, 8.1 Hz ), and 2.82 ( $2 \mathrm{H}, \mathrm{br} \mathrm{t}$ ) indicated the presence of a phenethyl moiety. Compound $\mathbf{1}$ thus belongs to the class of phenylethanoid natural products. The ${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{1}$, in addition to signals attributable to the phenethyl and the feruloyl groups, contained signals for 18 carbons corresponding to the carbohydrate moiety. The signals in the ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{1}$ for the anomeric protons at $\delta 5.19(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.4 \mathrm{~Hz}), 4.62(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.5$

Table 1. NMR Data of Compound 1, Poliumoside (8), and Martynoside (10)

|  | compound 1 |  | poliumoside (8) | martynoside (10) ${ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
|  | $\delta_{\mathrm{H}}(\mathrm{J}$ value in Hz$)$ | $\delta_{\text {c }}$ | $\delta_{\text {C }}$ | $\delta_{\text {c }}$ |
| aglycon |  |  |  |  |
| A-1 |  | 132.8 | 131.4 | 132.7 |
| A-2 | 6.73 (d, 2.1) | 117.1 | 117.1 | 117.0 |
| A-3 |  | 147.6 | 146.1 | 147.1 |
| A-4 |  | 147.3 | 144.7 | 147.2 |
| A-5 | 6.83 (d, 8.1) | 112.9 | 116.3 | 112.8 |
| A-6 | 6.69 (dd, 2.1, 8.1) | 121.2 | 121.3 | 121.1 |
| A-7 | 2.82 (dt) | 72.3 | 72.3 | 72.2 |
| A-8 |  | 36.6 | 36.5 | 36.5 |
| Caffeoyl- |  |  |  |  |
| C-1 |  | 127.7 | 127.6 | 127.5 |
| C-2 | 7.2 (d, 1.9) | 111.8 | 115.2 | 111.9 |
| C-3 |  | 149.4 | 146.8 | 150.4 |
| C-4 |  | 150.8 | 149.8 | 149.1 |
| C-5 | 6.81 (d, 8.2) | 116.5 | 116.5 | 116.6 |
| C-6 | 7.08 (dd, 1.9, 8.2) | 124.4 | 123.2 | 124.2 |
| C-7 | 7.66 (d. 15.9) | 148.0 | 148.0 | 147.8 |
| C-8 | 6.38 (d, 15.9) | 115.1 | 114.7 | 115.1 |
| -COOR |  | 168.0 | 168.0 | 168.3 |
| Glucosyl- |  |  |  |  |
| G-1 | 4.38 (d, 7.9) | 104.4 | 104.2 | 104.2 |
| G-2 |  | 76.2 | 76.2 | 76.2 |
| G-3 |  | 81.5 | 81.6 | 81.5 |
| G-4 |  | 70.4 | 70.6 | 70.6 |
| G-5 |  | 74.7 | 74.5 | 76.6 |
| G-6 |  | 67.6 | 67.5 | 62.4 |
| Rhamnosyl-1 |  |  |  |  |
| Rh'-1 | 5.19 (d, 1.7) | 103.0 | 103.1 | 103.0 |
| Rh'-2 |  | 72.3 | 72.4 | 72.3 |
| Rh'-3 |  | 72.1 | 72.1 | 72.0 |
| Rh'-4 |  | 73.9 | 73.9 | 73.8 |
| Rh'-5 |  | 69.9 | 70.4 | 70.4 |
| Rh'-6 |  | 18.4 | 18.4 | 18.4 |
| Rhamnosyl-2 |  |  |  |  |
| $\mathrm{Rh}^{\prime \prime}-1$ | 4.62 (d, 1.5) | 102.3 | 102.3 |  |
| Rh' -2 |  | 72.3 | 72.3 |  |
| Rh" -3 |  | 72.0 | 72.0 |  |
| Rh' -4 |  | 73.7 | 73.7 |  |
| Rh' -5 |  | 69.9 | 69.9 |  |
| Rh" -6 |  | 18.0 | 18.0 |  |
| $\mathrm{CH}_{3} \mathrm{O}$-(Caff-3) | 3.81 (s) | 56.5 |  | 56.5 |
| $\mathrm{CH}_{3} \mathrm{O}-(\mathrm{Ag}-4)$ | 3.88 (s) | 56.4 |  | 56.5 |

a Data from Miyase et al. 12c.
$\mathrm{Hz})$, and $4.38(1 \mathrm{H}, \mathrm{d}, 7.9 \mathrm{~Hz})$ and for the terminal methyl groups at $\delta 1.09(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.1 \mathrm{~Hz})$ and $1.19(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=$ 6.3 Hz ) indi cated that 1 contained two rhamnosyl units and one glucosyl unit.

Comparison of the NMR data of $\mathbf{1}$ with those of the known compound poliumoside (8) (Table 1) indicated that both carbon and proton chemical shifts of the sugar part of $\mathbf{1}$ were very similar to those of poliumoside (8). Compound 1 also had very similar ${ }^{13} \mathrm{C}$ NMR data to those of martynoside 10,12 after making adjustments for the fact that $\mathbf{1 0}$ has one rhamnose less than compound 1. A careful comparison of the ${ }^{13} \mathrm{C}$ NMR data of $\mathbf{1}$ with the data of $\mathbf{1 0}$ showed that the only significant differences occurred at $\mathrm{G}_{5}$ and $\mathrm{G}_{6}$ in 1, thus suggesting that the additional rhamnosylation in $\mathbf{1}$ took place at the $\mathrm{G}_{6}$ position of 10. ${ }^{12}$ The longrange couplings between $\mathrm{H}-\mathrm{G}_{4}\left(\delta_{\mathrm{H}} 4.99\right)$ and -COOR ( $\delta_{\mathrm{C}}$ 168.0) in the HMBC spectrum of 1 indicated that the feruloyl group was connected at $\mathrm{G}_{4}$ via an ester linkage, and the long-range couplings between $\mathrm{H}-\mathrm{Rh}_{1}\left(\delta_{\mathrm{H}} 5.19\right)$ and $\mathrm{C}-\mathrm{G}_{3}\left(\delta_{\mathrm{C}} 81.5\right)$ and between $\mathrm{H}-\mathrm{Rh}^{\prime}{ }_{1}\left(\delta_{\mathrm{H}} 4.62\right)$ and $\mathrm{C}-\mathrm{G}_{6}$ ( $\delta_{C} 67.6$ ) demonstrated that one rhamnosyl group was located at the $\mathrm{G}_{6}$ and the other at the $\mathrm{G}_{3}$ position. The correlation between $\mathrm{H}-\mathrm{G}_{1}\left(\delta_{\mathrm{H}} 4.38\right)$ and $\mathrm{C}-\alpha\left(\delta_{\mathrm{C}} 72.3\right)$ indicated the location of the $\alpha$-phenylethanyl glucoside moiety.


In a NOESY spectrum, NOE correlations observed between the $\mathrm{CH}_{3} \mathrm{O}-$ signal at $\delta_{\mathrm{H}} 3.81$ and the proton signal at $\delta_{H} 7.20(1 \mathrm{H}, \mathrm{d}, 2.1)$ showed this $\mathrm{CH}_{3} \mathrm{O}-$ group to be located at theC-3 position of the feruloyl moiety. Similarly, the other $\mathrm{CH}_{3} \mathrm{O}$ - group could be assigned to the C-4 position of the phenethyl group by the correlation between the $\mathrm{CH}_{3} \mathrm{O}$ - signal at $\delta_{\mathrm{H}} 3.88(3 \mathrm{H}, \mathrm{s})$ and proton signal at $\delta_{\mathrm{H}} 6.83(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.1 \mathrm{~Hz})$. Based on these data, 1 could be assigned as 2-(3-hydroxy-4-methoxyphenyl)-ethyl-O-( $\alpha$ -L-rhamnosyl)-(1 $\rightarrow 3$ )-O-( $\alpha-$ L-rhamnosyl)-(1 $\rightarrow 6$ )-4-O-E-feruloyl- $\beta$-D-glucopyranoside.

Compounds 2-9 were assigned as calceolarioside $A,{ }^{13}$ cal ceol arioside $B,{ }^{13}$ forsythiaside, ${ }^{14}$ plantainoside D, ${ }^{15}$ leucoceptoside, ${ }^{12}$ acteoside, $, 9,12,13,16$ poliumside, ${ }^{17}$ and plantarenaloside, ${ }^{17}$ respectively, by comparison of their ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, DQCOSY, HMQC, HMBC, and NOE SY spectra with literature data. They also showed the same HRFABMS, $[\alpha]_{D}$, UV, and IR data with values in the literature. Calceol ariosideA and plantarenal oside were al so identified by direct comparison with authentic samples. Plantainoside $D$ was isolated for the first time from $P$. Iinarioides.
All of the isolated compounds except compound 9 showed inhibitory activity against PKC $\alpha$, with I $\mathrm{C}_{50}$ values $(\mu \mathrm{M})$ of 125 (1), 0.6 (2), 4.6 (3), 1.9 (4), 14.8 (5), 19.0 (6), 9.3 (7) and 24.4 (8). The PKC $\alpha$ inhibitory bioactivities of 1-6 and 8 have not been reported previously in the literature.

## Experimental Section

General Experimental Procedures. Optical rotations were recorded with a Perkin-Elmer 241 Polarimeter. UV spectra were measured on a Beckman DU-50 instrument and IR spectra on a Nicolet Impact 400 spectrophotometer. NMR spectra were recorded in $\mathrm{CD}_{3} \mathrm{OD}$ on a Varian Unity 400 NMR instrument at 399.951 MHz for ${ }^{1} \mathrm{H}$ and 100.578 MHz for ${ }^{13} \mathrm{C}$, using standard Varian pulse sequences. Exact mass measurements were obtained at the Nebraska Center for Mass Spectrometry. Other conditions were as previously described. ${ }^{2}$

Plant Material. Stems, leaves, and fruit of Digitalis purpurea L. (Scrophulariaceae) were collected on Norfolk Island in May 1965 (PR-9503, B633363), and the whole plant of Penstemon Iinarioides Gray (Scrophulariaceae) was collected in New Mexico. in J une 1964 (PR-8513, B632569). Voucher specimens are on deposit in the Herbarium of the National Arboretum, Agricultural Research Service, U. S. D. A., Washington, DC. The dried samples were extracted with $\mathrm{CH}_{2-}$ $\mathrm{Cl}_{2}-\mathrm{MeOH}(1: 1)$ to give 2.52 g of D . purpurea extract as HEX 652 and 2.27 g of P . Iinarioide extract as HEX 1216.

Isolation of Phenylethanoids 1-5. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ extract HEX $652(2.52 \mathrm{~g})$ from D. purpurea was passed through a polyamide column ( 80 g , ICN Pharmaceuticals, Inc., Eschwege, Germany) to yield a detanninated fraction ( 2.16 g , PKC $\alpha \mathrm{IC}_{50}=78 \mu \mathrm{~g} / \mathrm{mL}$ ). After partition of the detanninated fraction between EtOAc and aqueous MeOH , followed by partition of the aqueous MeOH fraction between $\mathrm{H}_{2} \mathrm{O}$ and
$\mathrm{n}-\mathrm{BuOH}$, a bioactive $\mathrm{n}-\mathrm{BuOH}$ fraction ( $1.14 \mathrm{~g}, \mathrm{PKC}_{\alpha} \mathrm{IC}_{50}=12$ $\mu \mathrm{g} / \mathrm{mL}$ ) was obtained. This was subjected to col umn chromatography on Si gel ( 60 g ) with elution by $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (8:2:0.1) to give seven fractions after combination of similar components as determined by TLC. Purification of active compounds from the bioactive fractions was achieved by preparative TLC on RP-18 with $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(55: 45)$ followed by column chromatography on Si gel with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ( $8: 2: 0.1$ ). Fraction 2 yielded calceolarioside A ( $2,16.8 \mathrm{mg}$, $0.67 \%)$; fraction 3 gave calceolarioside B ( $3,8.8 \mathrm{mg}, 0.35 \%$ ); fraction 4 gave forsythiaside (4, 22.3 mg, $0.89 \%$ ); fraction 6 gave plantainoside D ( $5,8.0 \mathrm{mg}, 0.32 \%$ ) and the new natural product $\mathbf{1}(4.79 \mathrm{mg}, 0.19 \%)$. The structures of compounds 2-5 were assigned by ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, DQCOSY, HMQC, HMBC, and NOESY spectra, and by HRFABMS, $[\alpha]_{D}$, UV, and IR. Cal ceolarioside A was also identified by direct comparison with an authentic sample. ${ }^{13}$

Isolation of Phenylethanoids 6-9. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (1:1) extract from P. linarioides was partitioned between EtOAc and aqueous MeOH , and the active aqueous MeOH fraction was then partitioned between $\mathrm{H}_{2} \mathrm{O}$ and n-BuOH. The bioactive n -BuOH fraction ( $1.27 \mathrm{~g}, 56.1 \%$, with $\mathrm{IC}_{50}=17.1 \mu \mathrm{~g} /$ mL against $\mathrm{PK}(\alpha)$ was subjected to column chromatography on Si gel with the solvent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (6:1 and 4:1) to give 10 fractions, of which fractions 4, 6, and 8 showed $\mathrm{PKC} \mathrm{\alpha-}$ inhibitory activity. Leucosceptoside A ( $6,120 \mathrm{mg}, 5.2 \%$ ) was isolated from fraction 4 by further Si gel column chromatography with the sol vent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (6:1). Acteoside (7, 145 $\mathrm{mg}, 6.4 \%$ ) was purified by polyamide col umn chromatography of fraction 6 , and poliumoside ( $8,62.7 \mathrm{mg}, 2.8 \%$ ) and plantarenaloside ( $9,36.8 \mathrm{mg}, 1.6 \%$ ) were obtained by Si gel column chromatography of fraction 8 with the sol vent $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-$ $\mathrm{H}_{2} \mathrm{O}$
(8:2:0.1). The structures of compounds $\mathbf{6 - 8}$ were assigned by comparison of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, NOESY, HMQC and HMBC spectra, and UV, MS, and $[\alpha]_{D}$ values with literature data; all data were identical to those published. ${ }^{10,12,14,15,19}$ Plantarenaloside (9) was identified by direct comparison with an authentic sample (co-TLC, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR).


Compound 1: light yellow gum-like substance, $[\alpha]^{23}{ }_{D}-60.4^{\circ}$ (c $0.22, \mathrm{MeOH}$ ), UV (MeOH) $\lambda_{\text {max }}(\log \epsilon) 218$ (4.06), 235 sh (3.98), 289 (3.85), and 330 (4.05) nm; IR $v$ max (KBr) 3600$3100(\mathrm{OH}), 1720$ (conjugated COOR), $1600\left(>\mathrm{C}=\mathrm{C}<\right.$ ); ${ }^{1 \mathrm{H}}$ and ${ }^{13} \mathrm{C}$ NMR data see Table 1; FABMS m/z 805.3077 (calcd for $\mathrm{C}_{37} \mathrm{H}_{50} \mathrm{O}_{19} \mathrm{Li}, 805.3106$ ).

PKC $\alpha$-Inhibitory Bioassay. Bioassay for inhibition of PKC was carried out as described previously. ${ }^{1}$

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## References and Notes

(1) Lee, K. K.; Bahler, B. D.; H ofmann, G. A.; Mattern, M. R.; J ohnson, R. K.; Kingston, D. G. I. J . Nat. Prod. 1998, 61, 1407-1409.
(2) For previous papers in this series see (a) Wu, C.; Gunatilaka, A. A. L.; McCabe, F. L.; J ohnson, R. K.; Kingston, D. G. I. J. Nat. Prod. 1997, 60, 1281-1286. (b) Valente, L. M. M.; Gunatilaka, A. A. L.; Kingston, D. G. I.; Patitucci, M. L.; Pinto, A. C. J . Nat. Prod. 1997, 60, 478-481. (c) Wijeratne, E. M. K.; Gunatilaka, A. A. L.; Kingston, D. G. I.; Haltiwanger, R. C.; Eggleston, D. S. Tetrahedron 1995, 51, 7877-7882. (d)J ohnson, R. K.; Bartus, H. F.; Hofmann, G. A.; Bartus, J. O.; Mong, S.-M.; Faucette, L.; McCabe, F. L.; Chan, J . A.; Mirabelli, C. K. In: In Vitro and in Vivo Models for the Detection of New Antitumor Drugs; Hanka, L. J., Kondo, T., White, R. J., Eds.; Organizing Committee of the 14th International Congress of Chemotherapy: Kyoto, J apan, 1986; pp 15-26. (e) Gunatilaka, A. A. L.; Samaranayake, G.; Kingston, D. G. I.; H ofmann, G. A.; J ohnson, R. K. J. Nat. Prod. 1992, 55, 1648-1654. (f) Gunatilaka, A. A. L.; Kingston, D. G. I.; J ohnson, R. K. PureAppl. Chem. 1994, 66, 22192222.
(3) (a) Blackshear, P. J.; Nairn, A. C.; Kuo, J. F. FASEB J. 1988, 2, 29572969. (b) Basu, A. Pharmacol. Ther. 1993, 59, 257-280.
(4) (a) Tyler, V. E.; Brady, L. R.; Robbers, J. E. Pharmacognosy, 9th ed.; Lea and Febiger: Philadelphia, 1988; pp 164-169. (b) Stoll, A.; J ucker, E. In Moderne Methoden der Pflanzenanalyse; Paech, K., Tracey, M. V., Eds.; Springer-Verlag: Berlin, 1955; Band III, pp 205268. (c) Dean, F. M. Naturally Occurring Oxygen Ring Compounds, Butterworth: New York, 1963.
(5) (a) Paris, R. Compt. Rend. 1954, 238, 932-934. (b) Clerc, A.; Paris, R. Compt. Rend. Soc. Biol. 1940, 133, 46-48. (c) Brew, E. J. C.; Thomson, R. H. J. Chem. Soc. (C), 1971, 2007-2010.
(6) Matsumoto, M.; Koga, S.; Shoyama, Y.; Nishioka, I. Phytochemistry 1987, 26, 3225-3227.
(7) Abdel-K ader, M. S. Chemical Studies of Some Plants Belonging to the Families: Solanaceae, Zygophyllaceae, Asclepiadaceae, and Scrophulariaceae, Ph.D. Thesis, Alexandria University, Alexandria, Egypt, 1994; pp 32-73.
(8) Teborg, D.; J unior, P. Planta Med. 1989, 55, 474-476.
(9) (a) Gering-Ward, B.; J unior, P. Planta Med. 1989, 55, 75-78. (b) Gering-Ward, B.; J unior, P. Planta Med. 1989, 55, 75-78.
(10) (a) Lira-Rocha, A.; Diaz, R.; J imenez, C. J. Nat. Prod. 1987, 50, 331332. (b) Gering, B.; Wichtle, M. J . Nat. Prod. 1987, 50, 1048-1054.
(11) (a) J imenez, C.; Riguera, R. Nat. Prod. Rep. 1994, 11, 591-606. (b) Kimura Y.; Okuda, H.; Nishibe, S.; Arichi, S. Planta Med. 1998, 148153.
(12) (a) Gafner, S.; Wolfender, J .-L.; Nianga, M.; Hostettmann, K. Phytochemistry 1997, 44, 687-690. (b) Nishimura, H.; Sasaki, H.; Inagaki, N.; Chin, M.; Mitsuhashi, H. Phytochemistry 1991, 30, 965969. (c) Miyase, T.; Koizumi, A.; Ueno, A.; Noro, T.; K uroyanagi, M.; Fukushima, S.; Akiyama, Y.; Takemoto, T. Chem. Pharm. Bull. 1982, 30, 2732-2737.
(13) Nicoletti, M.; Galeffi, C.; Messana, I.; Garbarino, J. A.; Nyandat, E.; Marini-Bettolo, G. B. Gazz. Chim. Ital. 1986, 116, 431-433.
(14) (a) Endo, K.; Takahashi, K.; Hikino, H. Heterocycles 1981, 16, 13111314. (b) Nishiki, S.; Okabe, K.; Tsukamoto, H.; Sakushima, A.; Hisada, S. Chem. Pharm. Bull. 1982, 30, 1048-1050.
(15) Miyase, T.; Ishino, M.; Akahori, C.; Ueno, A.; Ohkawa, Y.; Tanizawa, H. Phytochemistry 1991, 30, 2015-2018.
(16) Andary, C.; Wylde, R.; Laffite, C.; Privat, G.; Winternitz, F. Phytochemistry 1982, 21, 1123-1127.
(17) Andary, C.; Wylde, R.; Heitz, A.; Rascol , J . P.; Laffite, C. Phytochemistry 1985, 24, 362-364.

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